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Programme and Abstracts

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Ce-CONTAINING POLYOXOTUNGSTATES: SYNTHESIS, FTIR SPECTROSCOPY, AND SURFACE MICROMORPHOLOGY

Oleksandra Yu. Mariichak^{1,2}, Tetiana O. Arzamastseva¹, Illia V. Kapitanov^{2,3}, Yevgen Karpichev^{2,3}, Georgii M. Rozantsev^{1,4}, Serhii V. Radio^{1,4}

¹Unit for Research, Research Group in Inorganic Chemistry, Vasyl' Stus Donetsk National University (in exile in Vinnytsia), vul. 600-richchia 21, Vinnytsia, UKRAINE

² Department of Chemistry, Chair of Green Chemistry, Tallinn University of Technology, Tallinn, ESTONIA
 ³ L.M. Litvinenko Institute of Physical Organic & Coal Chemistry NAS of Ukraine, Kyiv, UKRAINE
 ⁴ Faculty of Chemistry, Department of Inorganic Chemistry and Analytical Chemistry, Vasyl' Stus Donetsk National University (in exile in Vinnytsia), vul. 600-richchia 21-215, Vinnytsia, UKRAINE

o.marijchak@donnu.edu.ua

Polyoxometalates are a class of metal oxide clusters of early transition metals (V, Mo, W, Nb, etc.), and they show strong Brönsted acidity, fast reversible multi-electron redox transformations under mild conditions, and adjustable acid-base and redox properties over a wide range. Therefore, the use of POMs as acidic and redox-bifunctional catalysts in homogeneous and heterogeneous systems is their most popular and important application area. Thus, cerium decatungstate $[Ce^{IV}(W_5O_{18})_2]^{8-}$ modified with cetylpyridinium cations was found to be an active and selective catalyst for the oxidation of secondary alcohols to ketones with hydrogen peroxide [1].

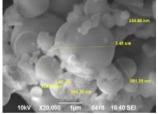
In present work we have studied the interactions of cerium (III) chloride with aqueous solutions of Na_2WO_4 at various acidity $Z = v(H^+)/v(WO_4^{2-}) = 0 - 1.50$. As known [2], during them isopoly tungstate anions of different composition $[H_{m-2k}W_nO_{4n-k}]^{(2n-m)-}$ are formed.

Presence of Ce(III) ions and other lanthanide ions can lead to the formation of heteropoly anions [3-4]:

 $Ce^{3+} + 10 WO_4^{2-} + 8 H^+ \leftrightarrows [Ce(W_5O_{18})_2]^{9-} + 4 H_2O, (Z = 0.80);$

 $2 \text{ Ce}^{3+} + 22 \text{ WO}_4^{2-} + 30 \text{ H}^+ \leftrightarrows [\text{Ce}_2(\text{H}_2\text{O})_{10}\text{W}_{22}\text{O}_{72}(\text{OH})_2]^{8-} + 4 \text{ H}_2\text{O}, (Z = 1.36).$

From aqua-acetone media (v/v=50/50) at Z=0.80 the sodium heteropoly decatungstocerate (III) $Na_9[Ce(W_5O_{18})_2]\cdot 28H_2O$ with Peacock-Weakley anion structure was synthesized. From acidified up to Z=0.80 aqueous solution of sodium tungstate without acetone adding the acid sodium heteropoly decatungstocerate (IV) $Na_6H_2[Ce(W_5O_{18})_2]\cdot 30H_2O$ with Peacock-Weakley anion structure was synthesized. Salts of these heteropoly compounds have different micromorphology (Fig. 1) and different positions of vibration maxima in the FTIR spectra (FTIR data for $Na_9[Ce(W_5O_{18})_2]\cdot 28H_2O$, cm⁻¹: 416, 485, 542, 575, 711, 787, 845, 955; FTIR data for $Na_6H_2[Ce^{IV}(W_5O_{18})_2]\cdot 30H_2O$, cm⁻¹: 436, 491, 553, 583, 677, 783, 828, 941).



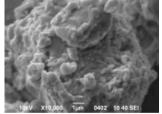


Fig. 1. SEM images of $Na_9[Ce(W_5O_{18})_2] \cdot 28H_2O$ powder surface (*left* – 20,000 times magnification), and $Na_6H_2[Ce(W_5O_{18})_2] \cdot 30H_2O$ (*right* – 10,000 times magnification)

From the solution with Z = 1.00 the cerium heptatungstate $Ce_2W_7O_{24} \cdot 20H_2O$ was synthesized, and by XRD it was shown that it thermal decomposition accompanied with crystallization of $Ce_4W_9O_{33}$ and WO_3 phases.

By the chemical analysis and FTIR spectroscopy it was shown that the $Ce_5[HW_7O_{24}]_3 \cdot 56H_2O$ was synthesized from solution with Z=1.17, and by XRD method it was shown that calcination of $Ce_5[HW_7O_{24}]_3 \cdot 56H_2O$ at $500^{\circ}C$ occurs with only WO₃ phase crystallization, and calcination at $700^{\circ}C$ accompanied with crystallization of $Ce_4W_9O_{33}$ phase. By the chemical analysis and FTIR spectroscopy it was shown that the cerium paratungstates $B-Ce_{10}[W_{12}O_{40}(OH)_2]_3 \cdot 94H_2O$ and $Na_{10}Ce_{20}[W_{12}O_{40}(OH)_2]_7 \cdot 180H_2O$ were synthesized from equilibria solutions with Z=1.17 and 1.29, respectively, and $Na_2Ce_2[Ce_2(H_2O)_{10}W_{22}O_{72}(OH)_2] \cdot 35H_2O$ with lacunar metatungstate-anions as ligands was synthesized from the solution at Z=1.364.

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